# Trifluralin 183



CAS Number: 1582-09-8

IUPAC name:  $\alpha$ ,  $\alpha$ ,  $\alpha$  -trifluoro-2,6-dinitro-N,N-dipropyl-p-toluidine

Chemical name: 2, 6-dinitro-N, N-dipropyl-4-(trifluoromethyl)benzenamine

Empirical formula: C<sub>13</sub>H<sub>16</sub>F<sub>3</sub>N<sub>3</sub>O<sub>4</sub>

RMM: 335.3

M.p.: 48.5 - 49°C

Solubility: In water 0.184 (pH 5), 0.221 (pH 7), 0.189 (pH 9) (all in mg/l)

In acetone, chloroform, acetonitrile, toluene, ethyl acetate >1000, methanol 33 - 40, hexane 50 - 67 (all in g/l,  $25^{\circ}$ C).

### Trifluralin Technical

#### 183/TC/M/-

**1. Sampling**. Take at least 100g

### 2. Identity tests

**2.1 HPLC**. Use the reversed phase HPLC method below. The relative retention time of the Trifluralin peak in the sample solution should not deviate by more than 1.5% from that of the calibration solution.

**2.2 Infrared**. Prepare potassium bromide discs for the technical sample and Trifluralin reference substance. A typical potassium bromide disc should contain a sample prepared in the 0.15-0.35% by weight range. Scan the discs from 4000 to 600 cm-1. The spectrum from the sample should not differ significantly from that of the reference substance.

# 3. Trifluralin

#### **OUTLINE OF METHOD**

Trifluralin is determined by reverse phase HPLC with DAD detector at 280nm and external standardization.

### REAGENTS

Trifluralin standard of known purity.

Acetonitrile: HPLC grade;

Water: Ultra-Pure;

Calibration solutions. Weigh in duplicate about 50 mg (to the nearest 0.1 mg) of trifluralin standard (s mg) into separate volumetric flasks (50 ml). Add acetonitrile (about 40ml) and place the flask in an ultrasonic bath for 5 min. Allow to cool to ambient temperature and fill to the mark with acetonitrile. Mix thoroughly (calibration solutions  $C_A$  and  $C_B$ ).

### **APPARATUS**

High performance liquid chromatograph equipped with a detector suitable for operation at 280 nm and an injection system capable of injecting 5ul.

Liquid chromatographic column stainless steel,  $150 \times 4.6 \text{ mm}$  (i.d.), packed with C<sub>18</sub>, 5 um, Eclipse Zorbax XDB-C18 or equivalent with the same selectivity.

### PROCEDURE

(a) HPLC conditions (typical) :

Column: Eclipse Zorbax XDB-C<sub>18</sub> 150 x 4.6 mm (i.d.), 5 um Mobile phase: Acetonitrile: water=77:23(V/V)Column temperature:  $25^{\circ}C$ Injection volume: 5ul Flow rate: 1.0ml/min Detector wavelength: 280nm Run time: 8.00 min Retention time: approximately 5.2 min

(b) Equilibration of the system

Pump sufficient mobile phase through the column to equilibrate the system. Inject 5  $\mu$ l portion of calibration solution C<sub>A</sub> until the retention time and peak area obtained from two consecutive injections deviate by less than 1.0%.

#### (c) Preparation of sample

Prepare sample solutions in duplicate for each sample. Weigh (to the nearest 0.1 mg) sufficient sample (w mg) to contain about 50mg of trifluralin (s mg) into a volumetric flask (50 ml). Add acetonitrile (about 40ml) and place the flask in an ultrasonic bath for 5 min. Allow to cool to ambient temperature and fill to the mark with acetonitrile. Mix thoroughly (sample solutions  $S_1$  and  $S_2$ ).

(d) Determination

Inject 5  $\mu$ l portion of calibration solution C<sub>B</sub>. The response factor for this solution should not deviate by more than 1.0% from that for calibration solution C<sub>A</sub>, otherwise prepare new calibration solutions. Inject in duplicate 5  $\mu$ l portions of each sample solution bracketing them by injections of the calibration solutions as follows: C<sub>A</sub>, S<sub>1</sub>, S<sub>1</sub>, C<sub>B</sub>, S<sub>2</sub>, S<sub>2</sub>, C<sub>A</sub>, and so on.

#### (e) Calculation

Determine the peak area of trifluralin and calculate the mean value of response factors from the calibration solutions bracketing the injections of the sample solutions and use this value for calculating the trifluralin content of the bracketed sample solutions. The trifluralin content is the mean value of two sample solutions.

$$f_i = \frac{s \times P}{H_s}$$

Trifluralin content = 
$$\frac{f \times H_w}{w}$$
 g/kg

where:

 $f_i \quad = individual \ response \ factor$ 

f = mean response factor

- $H_s$  = peak area of trifluralin in the calibration solution
- $H_w$  = peak area of trifluralin in the sample solution
- s = mass of trifluralin reference standard in the calibration solution (mg)

## TRIFLURALIN EMULSIFIABLE CONCENTRATES

#### 183/EC/M/-

**1. Sampling**. Take at least 100 ml.

#### 2. Identity test

- 2.1. HPLC As for Trifluralin technical 221/TC/M2.1
- 2.2 Infrared As for Trifluralin technical 221/TC/M 2.2

# 3. Trifluralin

As for Trifluralin technical 221/TC/M except:

(c) Preparation of sample. Prepare sample solutions in duplicate for each sample. Weigh (to the nearest 0.1 mg) sufficient sample (w mg) to contain about 50 mg of Trifluralin (s mg) into a volumetric flask (50 ml). Add acetonitrile (about 40ml) and place the flask in an ultrasonic bath

for 5 min. Allow to cool to ambient temperature and fill to the mark with acetonitrile. Mix thoroughly (sample solutions  $S_1$  and  $S_2$ ).



IR spectrum of Trifluralin



Typical HPLC-chromatogram of trifluralin standard



Typical HPLC-chromatogram of trifluralin TC



Typical HPLC-chromatogram of trifluralin EC